## Tamkang University Academic Year 111, 1st Semester Course Syllabus

| Course Title         | ADVANCED CHEMICAL ENGINEERING THERMODYNAMICS                                    | Instructor | YANG, YAN-LING  |
|----------------------|---|------------|---|
| Course Class         | TEDXM1A  MASTER'S PROGRAM, DEPARTMENT OF CHEMICAL AND MATERIALS ENGINEERING, 1A | Details    | <ul><li>◆ General Course</li><li>◆ Selective</li><li>◆ One Semester</li></ul> |
| Relevance<br>to SDGs | SDG8 Decent work and economic growth  |            |   |

#### Departmental Aim of Education

Education Objectives: Cultivation of chemical/materials engineering experts with professional knowledge and high research-and-development capability.

#### Subject Departmental core competences

- A. Possess the advanced knowledge of chemical/material engineering and to be able to use it. (ratio:40.00)
- B. Capable to plan and execute the chemical/material engineering projects.(ratio:5.00)
- C. Capable of writing professional papers.(ratio:5.00)
- D. Capable of creative thinking and solving problem independently.(ratio:30.00)
- E. Capable to coordinate and integrate interdisciplinary cooperation.(ratio:5.00)
- F. Possess global vision.(ratio:5.00)
- G. Qualified capability for leadership, management and planning.(ratio:5.00)
- H. Capable of self-learning and self-growth.(ratio:5.00)

### Subject Schoolwide essential virtues

- 1. A global perspective. (ratio:10.00)
- 2. Information literacy. (ratio:10.00)
- 3. A vision for the future. (ratio:10.00)
- 4. Moral integrity. (ratio:10.00)
- 5. Independent thinking. (ratio:30.00)
- 6. A cheerful attitude and healthy lifestyle. (ratio:10.00)
- 7. A spirit of teamwork and dedication. (ratio:10.00)
- 8. A sense of aesthetic appreciation. (ratio:10.00)

# Course Introduction

This course aims at graduate students. The purposes of this course are to review basic first and second laws of thermodynamics and introduce the concept of phase equilibrium and theory and application of solutions. That will help students understand thoroughly about the chemical thermodynamics and learn how to apply knowledge of that to the research theories.

# The correspondences between the course's instructional objectives and the cognitive, affective, and psychomotor objectives.

Differentiate the various objective methods among the cognitive, affective and psychomotor domains of the course's instructional objectives.

I. Cognitive: Emphasis upon the study of various kinds of knowledge in the cognition of the course's veracity, conception, procedures, outcomes, etc.

II. Affective: Emphasis upon the study of various kinds of knowledge in the course's appeal, morals, attitude, conviction, values, etc.

III.Psychomotor: Emphasis upon the study of the course's physical activity and technical manipulation.

| No.  |   |                              | objective methods |                     |   |  |  |
|------|---|------------------------------|-------------------|---------------------|---|--|--|
| 1    | Review laws o   | of thermo                    | Cognitive         |                     |   |  |  |
|      | Cultivate stud  |                              | Cognitive         |                     |   |  |  |
|      | The correspondences of teaching objectives: core competences, essential virtues, teaching methods, and assessment |                              |                   |                     |   |  |  |
| No.  | Core Competences  |                              | Essential Virtues | Teaching Methods    | Assessment                                  |  |  |
| 1    | ABCDEFGH  |                              | 12345678          | Lecture, Discussion | Testing, Report(including oral and written) |  |  |
| 2    | ABCDEFGH  |                              | 12345678          | Lecture, Discussion | Testing, Report(including oral and written) |  |  |
|      | Course Schedule   |                              |                   |                     |   |  |  |
| Week | Date  | Course Contents Note         |                   |                     |   |  |  |
| 1    | 111/09/05 ~<br>111/09/11  | Introduction                 |                   |                     |   |  |  |
| 2    | 111/09/12 ~<br>111/09/18  | Conservation of Energy (I)   |                   |                     |   |  |  |
| 3    | 111/09/19 ~<br>111/09/25  | Conservation of Energy (II)  |                   |                     |   |  |  |
| 4    | 111/09/26 ~<br>111/10/02  | Conservation of Energy (III) |                   |                     |   |  |  |

| 5                                   | 111/10/03 ~<br>111/10/09 | Entropy (I)  |  |  |  |
|-------------------------------------|--------------------------|--|--|--|--|
| 6                                   | 111/10/10 ~<br>111/10/16 | Entropy (II)   |  |  |  |
| 7                                   | 111/10/17 ~              | Entropy (III)  |  |  |  |
| 8                                   | 111/10/24 ~<br>111/10/30 | Industrial Applications (I)  |  |  |  |
| 9                                   | 111/10/31 ~<br>111/11/06 | Industrial Applications (II)   |  |  |  |
| 10                                  | 111/11/07 ~<br>111/11/13 | Midterm  |  |  |  |
| 11                                  | 111/11/14 ~<br>111/11/20 | Calculation of the Properties of Pure Fluids (I)   |  |  |  |
| 12                                  | 111/11/21 ~<br>111/11/27 | Calculation of the Properties of Pure Fluids (II)  |  |  |  |
| 13                                  | 111/11/28 ~<br>111/12/04 | Phase Behavior of Pure Fluids (I)  |  |  |  |
| 14                                  | 111/12/05 ~<br>111/12/11 | Phase Behavior of Pure Fluids (II)   |  |  |  |
| 15                                  | 111/12/12 ~<br>111/12/18 | Thermodynamic Properties of Mixtures (I)   |  |  |  |
| 16                                  | 111/12/19 ~<br>111/12/25 | Thermodynamic Properties of Mixtures (II)  |  |  |  |
| 17                                  | 111/12/26 ~<br>112/01/01 | Thermodynamic Properties of Mixtures (III)   |  |  |  |
| 18                                  | 112/01/02 ~<br>112/01/08 | Final  |  |  |  |
| Re                                  | quirement                |  |  |  |  |
| Teaching Facility                   |                          | Computer, Projector, Other (板書)  |  |  |  |
| Textbooks and<br>Teaching Materials |                          | Sandler, S. I.; "Chemical, Biochemical, and Engineering Thermodynamics", 4th Edition, John Wiley & Sons, 2006. |  |  |  |
| R                                   | References               |  |  |  |  |
| Number of<br>Assignment(s)          |                          | (Filled in by assignment instructor only)  |  |  |  |
| Grading<br>Policy                   |                          | <ul> <li>↑ Attendance:  %</li></ul>  |  |  |  |
|                                     |                          |  |  |  |  |

| Note | This syllabus may be uploaded at the website of Course Syllabus Management System at <a href="http://info.ais.tku.edu.tw/csp">http://info.ais.tku.edu.tw/csp</a> or through the link of Course Syllabus Upload posted on the home page of TKU Office of Academic Affairs at <a href="http://www.acad.tku.edu.tw/CS/main.php">http://www.acad.tku.edu.tw/CS/main.php</a> . |
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