## Tamkang University Academic Year 110, 1st Semester Course Syllabus

Course Title	ADVANCED CHEMICAL ENGINEERING THERMODYNAMICS Instructor YANG, YAN-LIN		YANG, YAN-LING			
Course Class	TEDXM1A <ul> <li>MASTER'S PROGRAM, DEPARTMENT OF</li> <li>CHEMICAL AND MATERIALS ENGINEERING, 1A</li> </ul> • General Course <ul> <li>Selective</li> <li>One Semester</li> </ul>					
Relevance to SDGs	SDG8 Decent work and economic growth					
Departmental Aim of Education						
Education Objectives: Cultivation of chemical/materials engineering experts with professional knowledge and high research-and-development capability.						
	Subject Departmental core competence	es				
<ul> <li>A. Possess the advanced knowledge of chemical/material engineering and to be able to use it. (ratio:70.00)</li> <li>D. Capable of creative thinking and solving problem independently.(ratio:30.00)</li> </ul>						
Subject Schoolwide essential virtues						
3. A vision for the future. (ratio:10.00) 5. Independent thinking. (ratio:90.00)						
Course Introduction	This course aims at graduate students. The purposes of this c basic first and second laws of thermodynamics and introduce equilibrium and theory and application of solutions. That will understand thoroughly about the chemical thermodynamics apply knowledge of that to the research theories.	course are to re e the concept of l help students and learn how	eview of phase ; v to			

The correspondences between the course's instructional objectives and the cognitive, affective, and psychomotor objectives. Differentiate the various objective methods among the cognitive, affective and psychomotor domains of the course's instructional objectives.						
<ul> <li>I. Cognitive : Emphasis upon the study of various kinds of knowledge in the cognition of the course's veracity, conception, procedures, outcomes, etc.</li> <li>II.Affective : Emphasis upon the study of various kinds of knowledge in the course's appeal, morals, attitude, conviction, values, etc.</li> <li>III.Psychomotor: Emphasis upon the study of the course's physical activity and technical manipulation.</li> </ul>						
No.	o. Teaching Objectives objective met					
1	Review laws of thermodynamics				Cognitive	
2	Cultivate students with professional knowledge of chemical thermodynamics				Cognitive	
	The correspondences of teaching objectives : core competences, essential virtues, teaching methods, and assessment					
No.	Core Competences		Essential Virtues	Teaching Methods	Assessment	
1	. AD		35	Lecture, Discussion	Testing, Report(including oral and written)	
2	2 AD		35	Lecture, Discussion	Testing, Report(including oral and written)	
	1			Course Schedule		
Week	Date Course Contents		rse Contents	Note		
1	110/09/22 ~ 110/09/28	Introdu	Introduction			
2	110/09/29~ 110/10/05	Conservation of Energy (I)				
3	110/10/06 ~ 110/10/12	Conser	Conservation of Energy (II)			
4	110/10/13~ 110/10/19	Conser	Conservation of Energy (III)			
5	110/10/20~ 110/10/26	Entropy (I)				
6	110/10/27 ~ 110/11/02	Entrop	Entropy (II)			
7	110/11/03 ~ 110/11/09	Entrop	Entropy (III)			
8	110/11/10~ 110/11/16	Industrial Applications (I)				
9	110/11/17 ~ 110/11/23	Industrial Applications (II)				
10	110/11/24~ 110/11/30 Midterm					

11	110/12/01~ 110/12/07	Calculation of the Properties of Pure Fluids (I)		
12	110/12/08~ 110/12/14	Calculation of the Properties of Pure Fluids (II)		
13	110/12/15~ 110/12/21	Phase Behavior of Pure Fluids (I)		
14	110/12/22 ~ 110/12/28	Phase Behavior of Pure Fluids (II)		
15	110/12/29~ 111/01/04	Thermodynamic Properties of Mixtures (I)		
16	111/01/05 ~ 111/01/11	Thermodynamic Properties of Mixtures (II)		
17	111/01/12 ~ 111/01/18	Thermodynamic Properties of Mixtures (III)		
18	111/01/19~ 111/01/25	Final		
Requirement				
Teaching Facility		Computer, Projector, Other (板書)		
Textbooks and Teaching Materials		Sandler, S. I.; "Chemical, Biochemical, and Engineering Thermodynamics", 4th Edition, John Wiley & Sons, 2006.		
References				
Number of Assignment(s)		(Filled in by assignment instructor only)		
Grading Policy		<ul> <li>Attendance: % ◆ Mark of Usual: % ◆ Midterm Exam: 50.0 %</li> <li>Final Exam: 50.0 %</li> <li>Other ⟨ ⟩: %</li> </ul>		
Note		This syllabus may be uploaded at the website of Course Syllabus Management System at <u>http://info.ais.tku.edu.tw/csp</u> or through the link of Course Syllabus Upload posted on the home page of TKU Office of Academic Affairs at <u>http://www.acad.tku.edu.tw/CS/main.php</u> . <b>※ Unauthorized photocopying is illegal. Using original textbooks is advised. It is a crime</b> <b>to improperly photocopy others' publications.</b>		

TEDXM1E1235 0A

Page:3/3 2021/7/19 11:18:38