Tamkang University Academic Year 109, 1st Semester Course Syllabus

Course Title	PHYSICAL AND CHEMICAL TREATMENT PROCESSES	Instructor	CHING-YU PENG		
Course Class	TEWXM1A MASTER'S PROGRAM, DEPARTMENT OF WATER RESOURCES AND ENVIRONMENTAL	Details	General CourseSelectiveOne Semester		
	'ENGINEERING, 1A Departmental Aim of Educ	ation			
 I. Cultivating students with capabilities of carrying out practical works or academic research related to water resources and environmental engineering. II. Cultivating students with capability of solving problems through researching, planning, 					
and management. Ⅲ. Cultivating students to become professional engineers with care in environment and professional ethics.					
-	IV. Preparing students with the capabilities of engaging in international engineering business, to adapt to globalization and social needs, and to expand their global perspectives.				
Subject Departmental core competences					
 C. Logical thinking, analysis, integration, problem-solving skills, engineering planning, design and implementation ability.(ratio:70.00) E. Capabilities of writing and presenting research report.(ratio:30.00) 					
Subject Schoolwide essential virtues					
2. Informa	2. Information literacy. (ratio:30.00)				
4. Moral ir	4. Moral integrity. (ratio:70.00)				
	Through journal papers reading, understand and learn how tenvironmental physical and chemical treatment processes for removal.				
Course Introduction					

The correspondences between the course's instructional objectives and the cognitive, affective, and psychomotor objectives.

Differentiate the various objective methods among the cognitive, affective and psychomotor domains of the course's instructional objectives.

I. Cognitive: Emphasis upon the study of various kinds of knowledge in the cognition of the course's veracity, conception, procedures, outcomes, etc.

II. Affective: Emphasis upon the study of various kinds of knowledge in the course's appeal, morals, attitude, conviction, values, etc.

III.Psychomotor: Emphasis upon the study of the course's physical activity and technical manipulation.

tive methods					
Understand the theory and mechanisms governing physical and Cognitive chemical treatment processes					
Understand and apply the theory underlying each process Cognitive					
The correspondences of teaching objectives : core competences, essential virtues, teaching methods, and assessment					
sessment					
uding oral and					
Course Schedule					
Note					
Ion Exchange (I)					
Ion Exchange (II)					
Ion Exchange (III)					
Granular Filtration (I)					
Granular Filtration (II)					
Secondary treatment (I)					
Secondary treatment (II)					
Midterm Exam					

11	109/11/23 ~ 109/11/29	Secondary Treatment (III)		
12	109/11/30 ~ 109/12/06	Carbon Adsorption (I)		
13	109/12/07 ~ 109/12/13	Carbon Adsorption (II)		
14	109/12/14 ~ 109/12/20	Carbon Adsorption (III)		
15	109/12/21 ~ 109/12/27	Removal of selected constituents (I)		
16	109/12/28 ~ 110/01/03	Removal of selected constituents (II)		
17	110/01/04 ~ 110/01/10	Removal of selected constituents (III)		
18	110/01/11 ~ 110/01/17	Final Exam		
Requirement		There will be paper presentation, and journal review. Missed paper presentation or journal review counts as a zero.		
Tea	eaching Facility Computer, Projector			
Textbooks and Teaching Materials		Water and Wastewater Engineering: Design Principles & Practice, Davis, 2011		
Deferences		Selected papers Water Treatment Principles and Design, 2nd ed. by Montgomery Watson Harza		
Number of Assignment(s) 2 (Filled in by assignment instructor only)		2 (Filled in by assignment instructor only)		
Grading Policy Attendance: 10.0 % ◆ Mark of Usual: % ◆ Midterm Exam: 20.0 % Final Exam: 20.0 % Other ⟨Oral Presentation⟩: 50.0 %		◆ Final Exam: 20.0 %		
	This syllabus may be uploaded at the website of Course Syllabus Management System at http://info.ais.tku.edu.tw/csp or through the link of Course Syllabus Upload posted on the home page of TKU Office of Academic Affairs at http://www.acad.tku.edu.tw/CS/main.php . ** Unauthorized photocopying is illegal. Using original textbooks is advised. It is a crime to improperly photocopy others' publications.			

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