## Tamkang University Academic Year 108, 2nd Semester Course Syllabus

Course Title	AQUATIC CHEMISTRY	Instructor	LI, CHI-WANG					
Course Class	TEWXM1A MASTER'S PROGRAM, DEPARTMENT OF WATER RESOURCES AND ENVIRONMENTAL	Details	<ul> <li>General Course</li> <li>Selective</li> <li>One Semester</li> </ul>					
	ENGINEERING, 1A Departmental Aim of Education							
I. Cultivating students with capabilities of carrying out practical works or academic research related to water resources and environmental engineering.								
	<ul> <li>I. Cultivating students with capability of solving problems through researching, planning, and management.</li> </ul>							
III. Cultivating students to become professional engineers with care in environment and professional ethics.								
IV. Preparing students with the capabilities of engaging in international engineering business, to adapt to globalization and social needs, and to expand their global perspectives.								
	Subject Departmental core competence	es						
<ul> <li>B. Capabilities of planning and conducting experiments, analyzing and explaining experimental data, applying information tool, and collecting and compiling data. (ratio:40.00)</li> <li>C. Logical thinking, analysis, integration, problem-solving skills, engineering planning, design and implementation ability.(ratio:40.00)</li> <li>D. Skill of using professional foreign language and global perspective.(ratio:20.00)</li> </ul>								
Subject Schoolwide essential virtues								
1. A global perspective. (ratio:50.00)								
5. Indeper	ndent thinking. (ratio:50.00)							
Course Introduction	Principles of chemical equilibrium related to the treatment of wastewater are introduced. The effects of chemical interactio industrial waste effluents on natural water system are discuss	n of domestic	and					

	The correspondences between the course's instructional objectives and the cognitive, affective, and psychomotor objectives.						
Differentiate the various objective methods among the cognitive, affective and psychomotor domains of the course's instructional objectives.							
II.A	the ffective : Emp mor sychomotor:	course's bhasis up als, attitu	veracity, conception, pr on the study of various ude, conviction, values, is upon the study of the	s kinds of knowledge in the cognition of ocedures, outcomes, etc. kinds of knowledge in the course's appea etc. • course's physical activity and technical	al,		
No.	Teaching Objectives objective method						
1	Understand the principles of chemical equilibrium and application Cognitive				Cognitive		
	The correspondences of teaching objectives : core competences, essential virtues, teaching methods, and assessment						
No.	Core Competences		Essential Virtues	Teaching Methods	Assessment		
1	BCD		15	Lecture, Discussion	Testing, Study Assignments		
		I		Course Schedule			
Week	Date	Cours		rse Contents	Note		
1	109/03/02 ~ 109/03/08	Introduction, Basic concept					
2	109/03/09~ 109/03/15	Fundamentals of thermodynamics					
3	109/03/16~ 109/03/22	Acid ar	Acid and base (I)				
4	109/03/23~ 109/03/29	Acid ar	Acid and base (II)				
5	109/03/30~ 109/04/05	Acid ar	Acid and base (III)				
6	109/04/06~ 109/04/12	Titratio	Titration and buffer				
7	109/04/13~ 109/04/19	Titratio	Titration and buffer				
8	109/04/20~ 109/04/26	Softwa	Software for solving Chemical equilibrium				
9	109/04/27 ~ 109/05/03	Midter	Midterm Exam Week				
10	109/05/04 ~ 109/05/10	Softwa	Software for solving Chemical equilibrium				
11	109/05/11~ 109/05/17	Gas/liquid equilibrium					
12	109/05/18~ 109/05/24	Gas/liquid equilibrium					
13	109/05/25~ 109/05/31	~ Gas/liquid equilibrium					

14	109/06/01~ 109/06/07	Metals		
15	109/06/08~ 109/06/14	Metals		
16	109/06/15~ 109/06/21	Metals		
17	109/06/22 ~ 109/06/28	Final Exam Week		
18	109/06/29 ~ 109/07/05	Supplementary teaching: Oxidation/reduction		
Requirement				
Teaching Facility		Computer		
Textbooks and Teaching Materials		Water Chemistry, Mark M. Benjamin.		
References				
Number of Assignment(s)		10 (Filled in by assignment instructor only)		
Grading Policy		<ul> <li>♦ Attendance: % ♦ Mark of Usual: % ♦ Midterm Exam: 20.0 %</li> <li>♦ Final Exam: 20.0 %</li> <li>♦ Other ⟨HW, two tests (40%)⟩: 60.0 %</li> </ul>		
	This syllabus may be uploaded at the website of Course Syllabus Management System at <a href="http://info.ais.tku.edu.tw/csp">http://info.ais.tku.edu.tw/csp</a> or through the link of Course Syllabus Upload posted on the         Note       home page of TKU Office of Academic Affairs at <a href="http://www.acad.tku.edu.tw/CS/main.php">http://www.acad.tku.edu.tw/CS/main.php</a> .         Wote <b>Winauthorized photocopying is illegal. Using original textbooks is advised. It is a control to improperly photocopy others' publications.</b>			

TEWXM1E1162 0A

Page:3/3 2020/2/28 16:19:59